

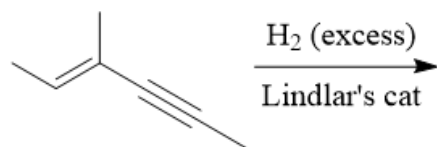
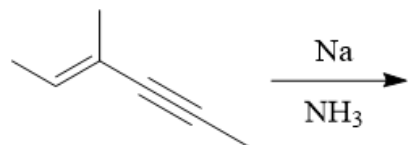
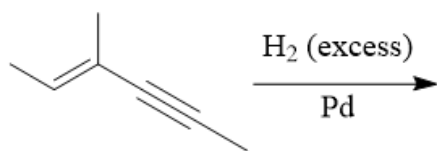
Dr. Starkey, CHM 3140 Organic Chem. I, Cal Poly Pomona  
Chapter 9 Alkyne Reactions, Part 1 – [Practice Problems](#)

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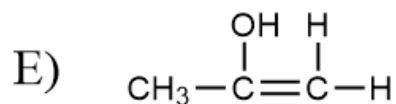
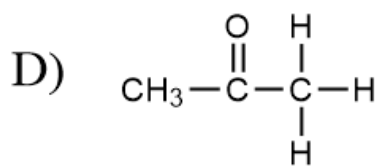
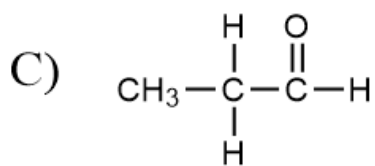
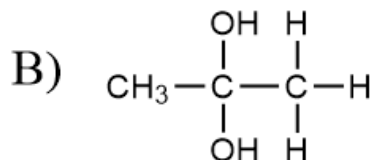
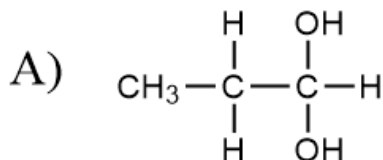
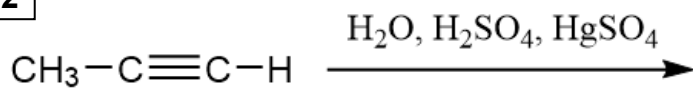


Predict the major products for the following reactions.

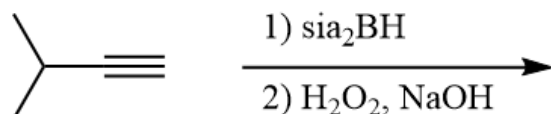
1



2



3



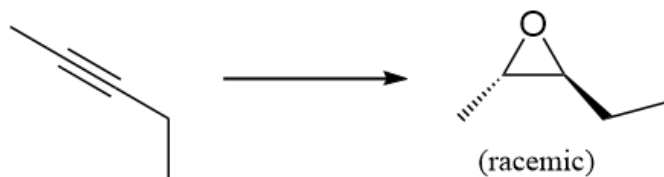
4



5



6



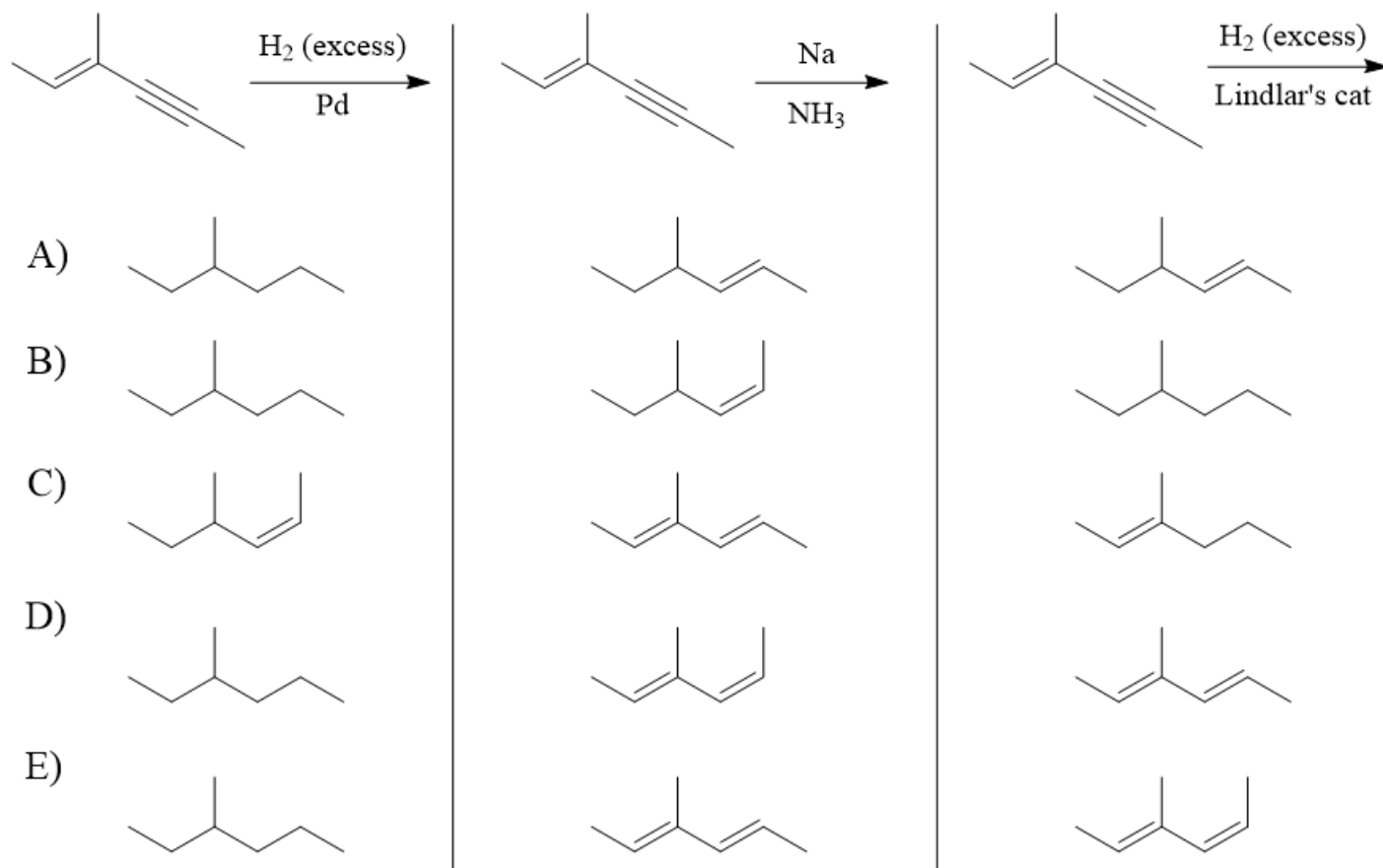
7

Which of the following is the major product of the reaction shown below?



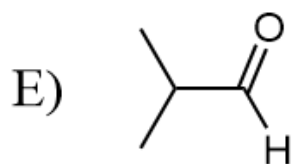
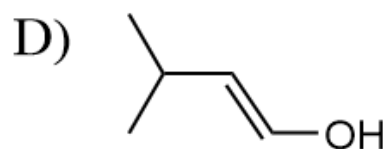
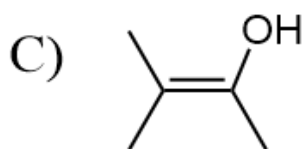
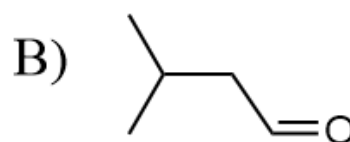
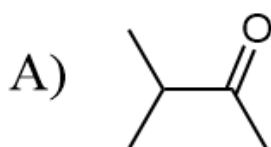
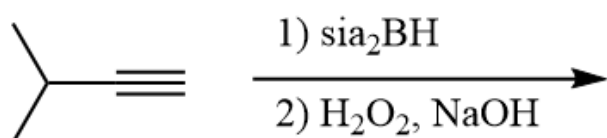
Predict the major products for the following reactions.

1



3

Predict the major product.

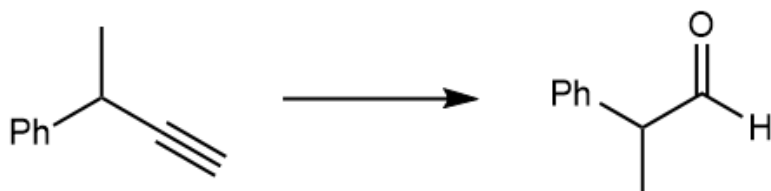


4 Which is the best set of reagents to accomplish the following transformation?



- A) 1)  $\text{H}_2$ , Lindlar's cat.  
2)  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$   
 $\text{HgSO}_4$
- B) 1)  $\text{H}_2$ , Lindlar's cat.  
2) 9-BBN  
3)  $\text{H}_2\text{O}_2$ , NaOH
- C) 1) 9-BBN  
2)  $\text{H}_2\text{O}_2$ , NaOH
- D)  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$   
 $\text{HgSO}_4$
- E) 1)  $\text{H}_2$ , Lindlar's cat.  
2)  $\text{O}_3$   
3) DMS

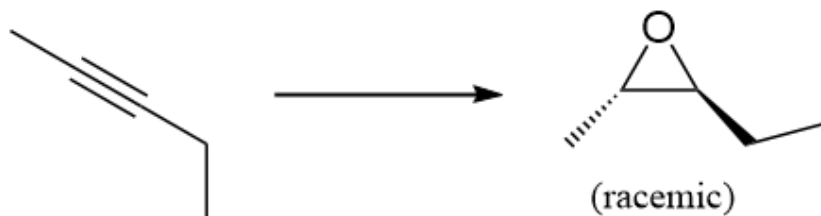
5 Which is the best set of reagents to accomplish the following transformation?



- A) 1)  $\text{H}_2$ , Lindlar's cat.  
2)  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$   
 $\text{HgSO}_4$
- B) 1)  $\text{H}_2$ , Lindlar's cat.  
2) 9-BBN  
3)  $\text{H}_2\text{O}_2$ , NaOH
- C) 1) 9-BBN  
2)  $\text{H}_2\text{O}_2$ , NaOH
- D)  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$   
 $\text{HgSO}_4$
- E) 1)  $\text{H}_2$ , Lindlar's cat.  
2)  $\text{O}_3$   
3) DMS

6

Which reagents would be best to achieve the following synthesis?



A) 1) Na, NH<sub>3</sub>  
2) cat. OsO<sub>4</sub>, NMO

C) 1) H<sub>2</sub>, Lindlar's cat.  
2) cat. OsO<sub>4</sub>, NMO

B) 1) H<sub>2</sub>, Lindlar's cat.  
2) MCPBA

D) 1) Na, NH<sub>3</sub>  
2) RCO<sub>3</sub>H

7

(from the Chemistry GRE practice test)

Which of the following is the major product of the reaction shown below?



A) NaCH<sub>2</sub>CH<sub>2</sub>C≡CH

D) 
$$\begin{array}{c} \text{NH}_2 \\ | \\ \text{CH}_3\text{CH}_2\text{C}=\text{CHNa} \end{array}$$

B) CH<sub>3</sub>CH<sub>2</sub>C≡CNH<sub>2</sub>

E) 
$$\begin{array}{c} \text{NH}_2 \\ | \\ \text{CH}_3\text{CH}_2\text{C}=\text{CH}_2 \end{array}$$

C) CH<sub>3</sub>CH<sub>2</sub>C≡CNa