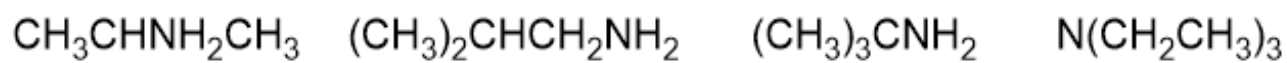


Organic Chemistry II CHM 3150  
Dr. Laurie S. Starkey, Cal Poly Pomona  
Chapter 22, Amines – [Practice Problems](#)

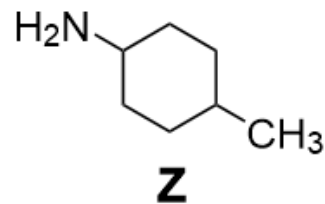
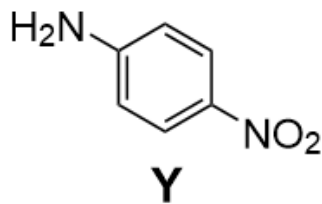
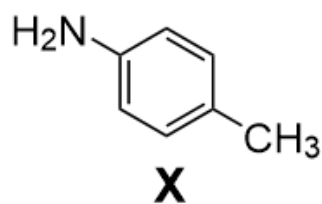
For clicker question voting, go to:  
<https://pollev.com/lauriestarke263>



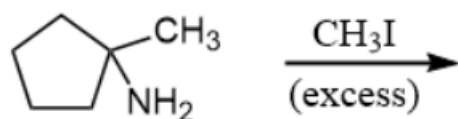
1 Provide both common and IUPAC names for each compound:



2 Which is the strongest base? Which is the weakest base? Explain briefly.

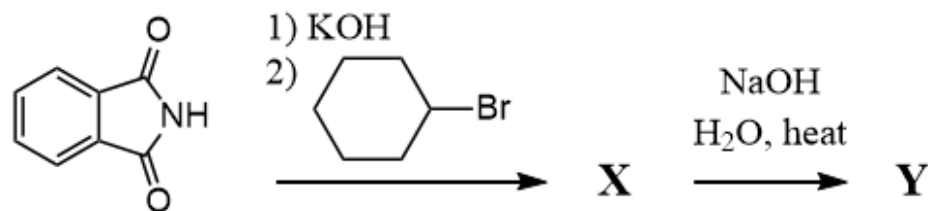


3 Predict the major product.

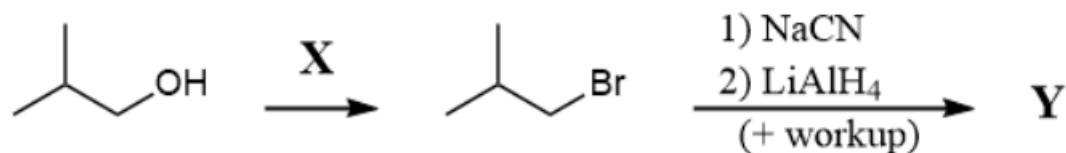


4

Predict the products **X** and **Y** and provide mechanisms for their formation.

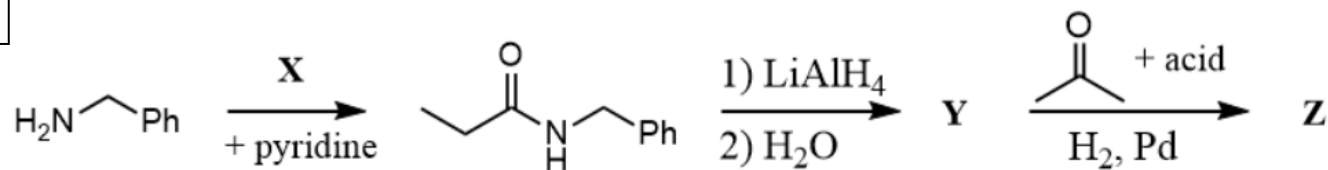


5 Identify the best reagent(s) **X** and the major product **Y**.

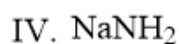
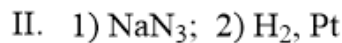
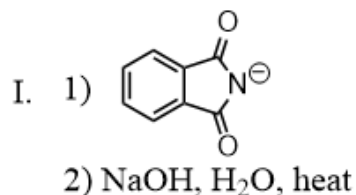
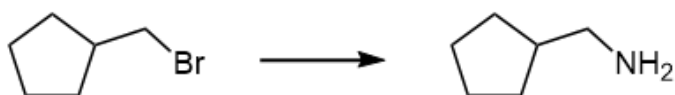


Provide the missing reagent **X** and predict the major products **Y** and **Z**.

6

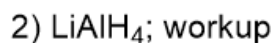
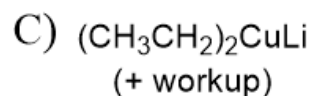
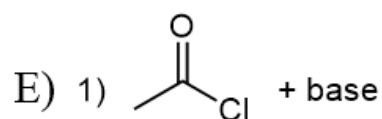
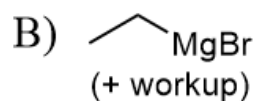
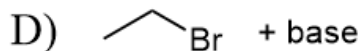
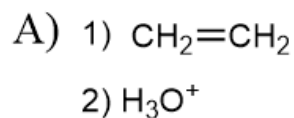
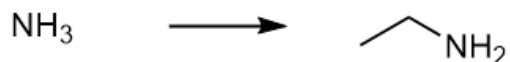


7 Which of the following reagents are suitable for the given transformation?



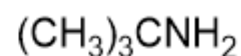
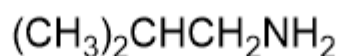
8

Of the following reagents, which is BEST for the given transformation?



1

Match the given compound with its correct common name.



A) isopropyl amine

*n*-butyl amine

isobutyl amine

B) *n*-propyl amine*tert*-butyl amine*sec*-butyl amine

C) isopropyl amine

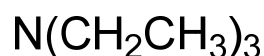
isobutyl amine

*tert*-butyl amineD) *n*-propyl amine*n*-butyl amine*tert*-butyl amine

E) isopropyl amine

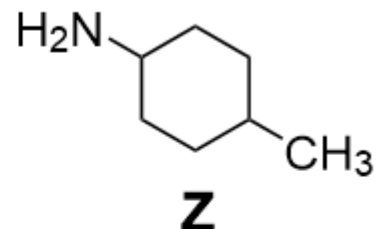
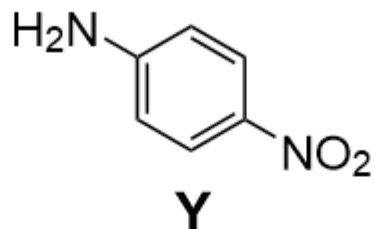
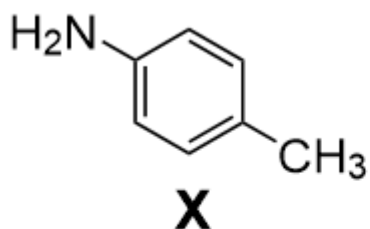
*tert*-butyl amine

isobutyl amine



2

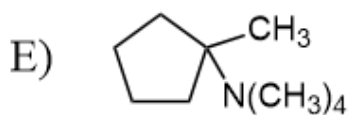
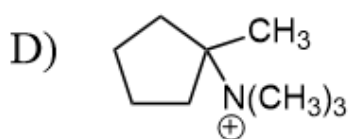
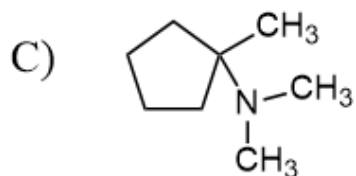
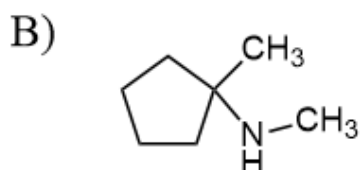
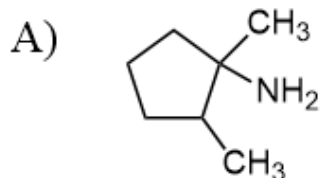
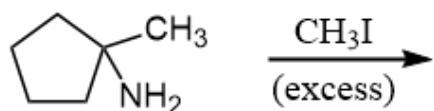
Which is the strongest base? Explain briefly.



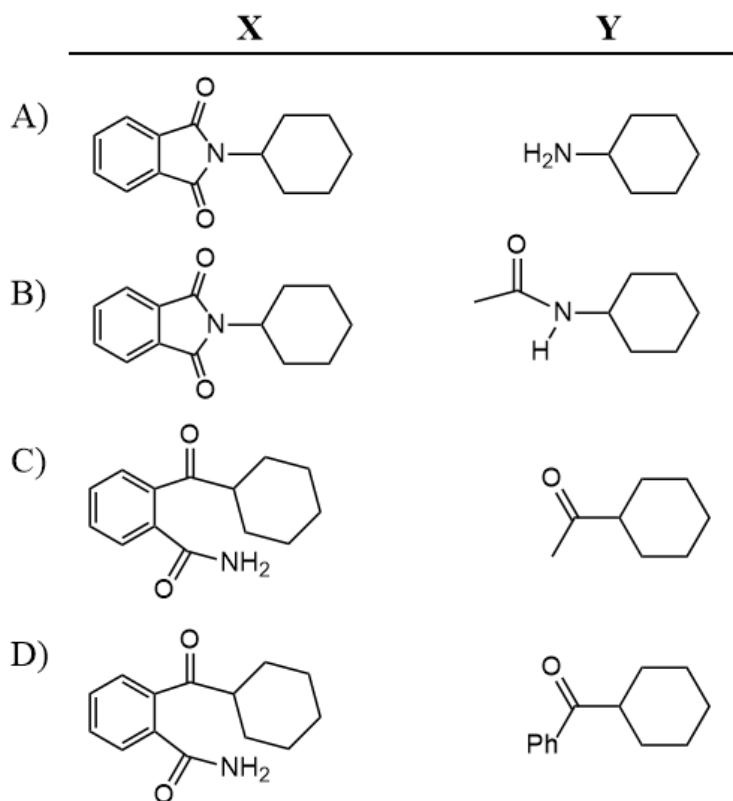
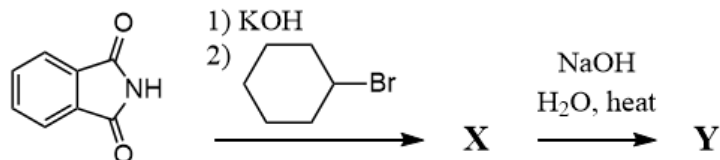
- A) **Y** is the strongest base because **Y** is destabilized by the EWG.
- B) **Y** is the strongest base because its conj. acid is stabilized by  $\text{NO}_2$ .
- C) **Z** is the strongest base because it has the least stable conj. acid.
- D) **Z** is the strongest base because the lone pair on N is localized.
- E) It's impossible to predict base strength without  $\text{pK}_a/\text{pK}_b$  data.

3

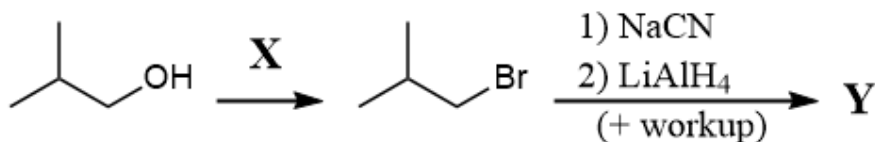
Predict the major product.



4



5 Identify the best reagent(s) and the major product.



reagent(s) **X**

product **Y**

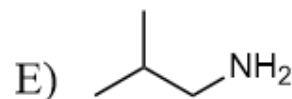
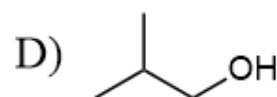
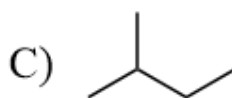
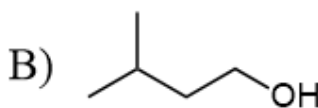
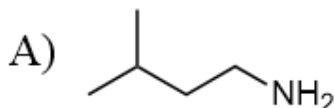
A) HBr

B) HBr, ROOR

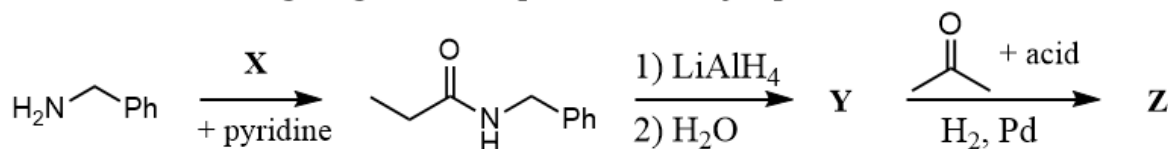
C) PBr<sub>3</sub>

D) Br<sub>2</sub>, hv

E) Br<sub>2</sub>, H<sub>2</sub>O

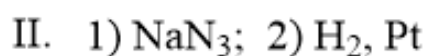
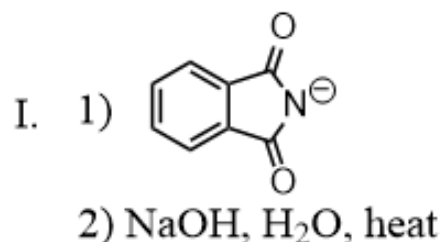
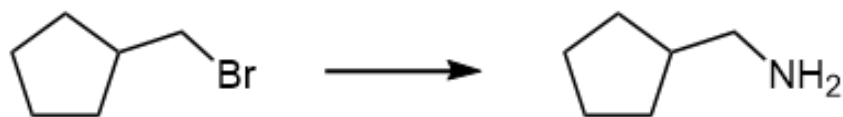


6 Provide the missing reagent **X** and predict the major products **Y** and **Z**.



	<b>X</b>	<b>Y</b>	<b>Z</b>
A)			
B)			
C)			
D)			

7 Which of the following reagents are suitable for the given transformation?



A) I and II only

B) II and IV only

C) I, II and IV only

D) I, II, III and IV

8

Of the following reagents, which is BEST for the given transformation?

