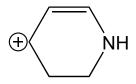
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4A. (6 pts) For the Lewis structure below, **supply any missing lone pairs of electrons** and draw ONE additional, significant resonance forms. **Rank** each of the resonance contributors in order of importance and **explain** your answers. (If any are equivalent, be sure to indicate so in your answer.)

$$N \equiv C - CH - CH_3$$

4B. (9 pts) For the Lewis structure below, **supply any missing lone pairs of electrons** and draw TWO additional resonance forms that are significant. **Rank** each of the three resonance contributors in order of importance and **explain** your answers. (If any are equivalent, be sure to indicate so in your answer.)



4C. (10 points) Which of the following compounds, when pure, can form hydrogen bonds? (write **yes or no** in the box; no explanation is needed.

Also, draw each compound as a bond-line drawing.

	$CH_3CH_2NHCH(CH_3)_2$		$(CH_3CH_2CH_2)_2O$			CH <sub>3</sub> COCH <sub>3</sub>		
	H-bond?		H-bond?			H-bond?		
bond-line drawing			bond-line drawing			bond-line drawing		