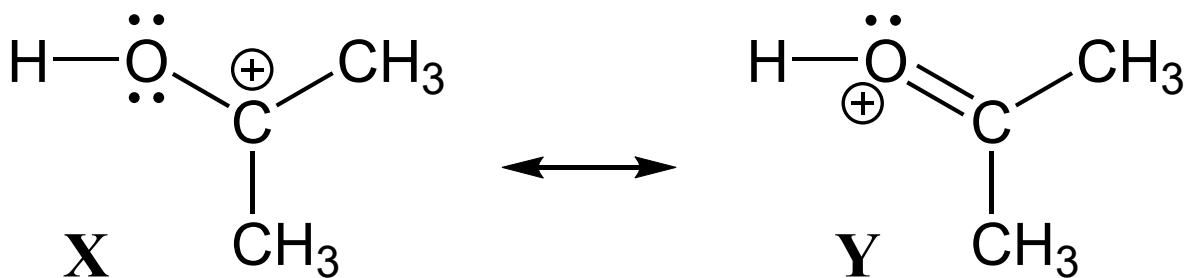


Which resonance structure contributes more to the resonance hybrid? Explain briefly.



- A) **X** because carbon better handles positive charge (C less electronegative)
- B) **Y** because oxygen better handles positive charge (O more electronegative)
- C) **Y** because oxygen better handles positive charge (O is larger).
- D) **X** because **Y** is missing an octet.
- E) **Y** because **X** is missing an octet.