

Draw the conjugate acid or conjugate base, as directed, for each.

	conjugate base of	conjugate acid of	conjugate acid of
	$\text{NH}_3$	$\text{H}_2\text{O}$	$\text{H}:\ominus$
A)	$\text{NH}_2^\ominus$	$\text{H}_3\text{O}^\oplus$	$\text{H}_2\text{O}$
B)	$\text{NH}_4^\oplus$	$\text{OH}^\ominus$	$\text{H}_2\text{O}$
C)	$\text{NH}_2^\ominus$	$\text{OH}^\ominus$	$\text{H}_2$
D)	$\text{NH}_2^\ominus$	$\text{H}_3\text{O}^\oplus$	$\text{H}_2$
E)	$\text{NH}_4^\oplus$	$\text{OH}^\ominus$	$\text{H}_2$