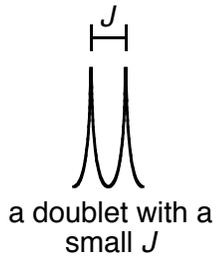


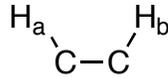
# Dr. Laurie S. Starkey, Cal Poly Pomona - NMR Spectroscopy: Spin-Spin Coupling

The magnitude of the coupling between two neighboring protons is determined by their spatial relationship. When considering the dihedral angle between hydrogens (as observed via a Newman projection), the maximum coupling constant ( $J$ ) occurs at  $0^\circ$  and  $180^\circ$  (eclipsed and anti protons, respectively), and is at a minimum when they are at  $90^\circ$  (orthogonal protons). Spin-spin coupling can occur through 2, 3 or even 4 bonds and can be generally classified as small ( $<3$  Hz), medium ( $\sim 7$  Hz) or large ( $>10$  Hz). Not all protons are split by neighboring protons. Protons on N or O are acidic and are exchangeable. As a result, these signals usually do not couple with neighboring protons (**OH and NH are typically broad singlets**). Also, since protons on carbonyls have very small coupling constants, **aldehyde H's typically appear as singlets**.

## General coupling constants (called "J values")



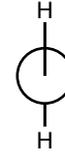
vicinal (3-bond)



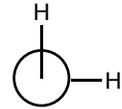
$J_{ab} \sim 7$  Hz  
(freely rotating, so no fixed dihedral angle)



$0^\circ$  dihedral angle  
large coupling  
 $J_{ab} \sim 2-14$  Hz

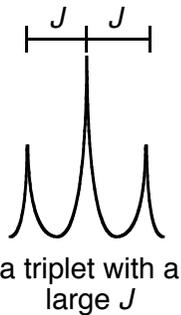


$180^\circ$  dihedral  
large coupling  
 $J_{ab} \sim 4-16$  Hz

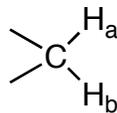


$90^\circ$  dihedral  
small coupling  
 $J_{ab} \sim 0-2$  Hz

(for fixed dihedral angles - no free rotation)



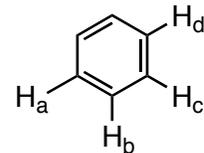
geminal (2-bond)



$J_{ab} \sim 10-15$  Hz

*note: only nonequivalent hydrogens will experience splitting, so this geminal coupling is for diastereotopic H's only (i.e., those near a chiral center).*

longer-range coupling is also possible (very small)



ortho  $J_{ab} \sim 7-10$  Hz  
meta  $J_{ac} \sim 2-3$  Hz  
para  $J_{ad} \sim 0-1$  Hz

geminal (alkene)

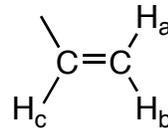
$J_{ab} \sim 1-2$  Hz

trans (alkene)

$J_{ac} \sim 11-18$  Hz

cis (alkene)

$J_{bc} \sim 6-15$  Hz



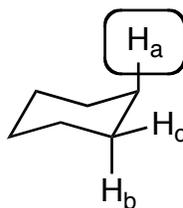
*note: gem coupling in an alkene ( $sp^2$   $CH_2$ ) is much smaller than for an alkane ( $sp^3$   $CH_2$ ):  $\sim 1$  vs.  $\sim 12$  Hz!*

cyclohexane (ax-ax)

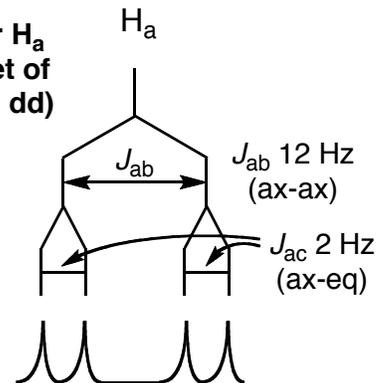
$J_{ab} \sim 10-13$  Hz  
( $180^\circ$  dihedral)

cyclohexane (ax-eq) or (eq-eq)

$J_{ac} \sim 2-5$  Hz  
( $60^\circ$  dihedral)



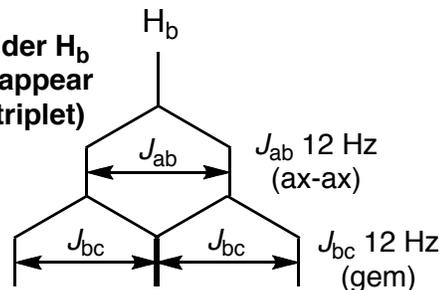
consider  $H_a$   
(a doublet of doublets, dd)



a doublet of doublets, dd  
(four peaks but not a quartet - all equal heights/integration, and not even spacing between peaks)

*note:  $H_b$  (a doublet of doublets, dd) may appear as an apparent triplet if the large gem coupling ( $J_{bc}$ ) is about the same as the large ax-ax coupling ( $J_{ab}$ ). i.e., if J values are equal, then two neighbors result in a triplet ( $n+1$  rule).*

consider  $H_b$   
(may appear as a triplet)



this dd appears as an apparent triplet, t (three peaks with even spacing and 1:2:1 peak height ratio)