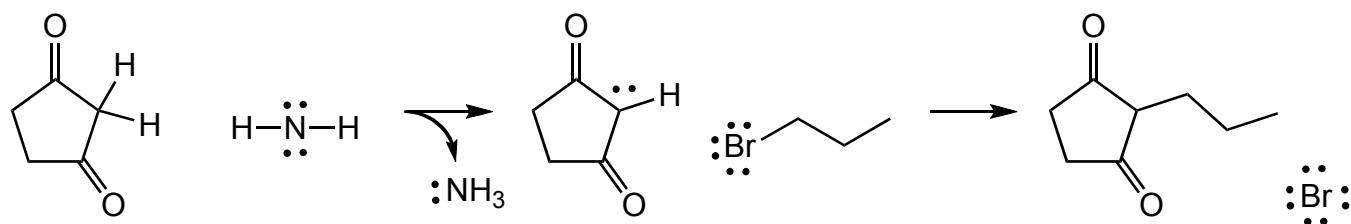
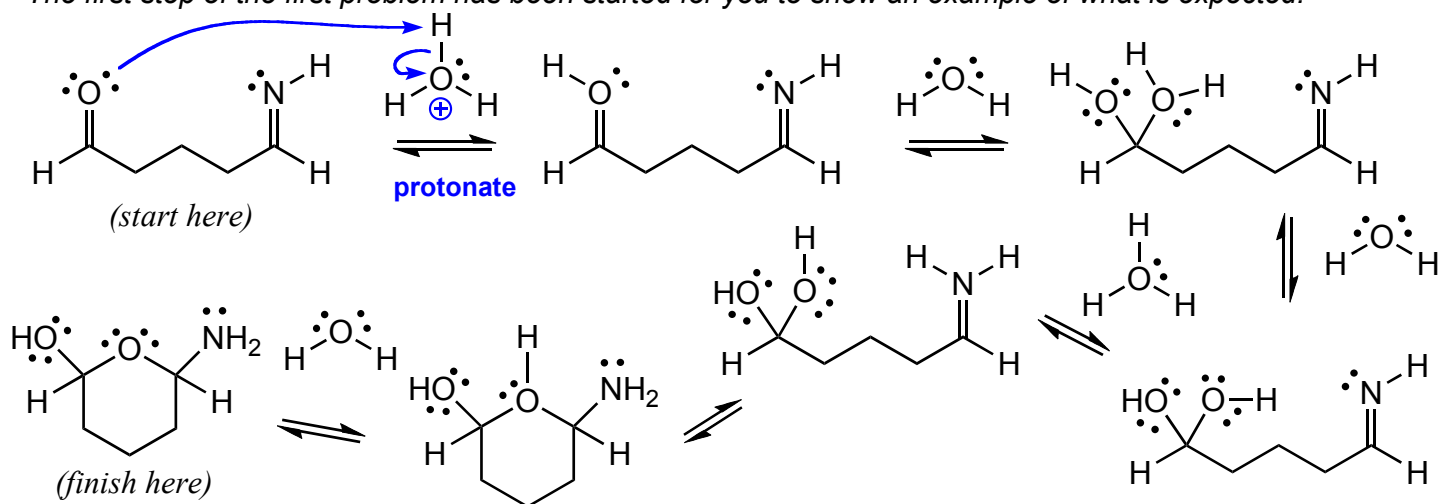
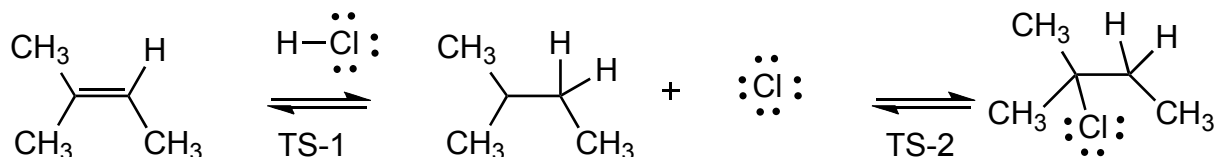
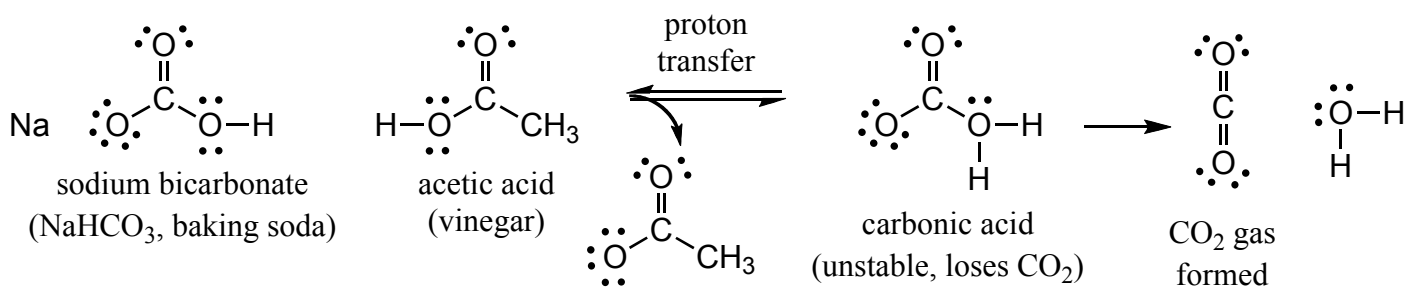


Formal Charges, Curved Arrows & Transition States

Add in any **missing formal charges**. Draw **curved arrows** for each step of the following mechanisms. Identify any **proton-transfer** (acid-base) steps and label the step as "protonate" or "deprotonate" to describe what is happening to the organic compound involved - if it accepts a H^+ it's a protonation step ("protonate"). *The first step of the first problem has been started for you to show an example of what is expected.*



homemade "volcano" or chemistry experiment in a bag (www.youtube.com/ChemistryConnected)



Draw the **structures of the transition states TS-1 and TS-2** for the above two-step mechanism.